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Synthesis of phosphodiester-type nicotinamide adenine dinucleotide analogs

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ABSTRACT

Fourteen phosphodiester-type β -nicotinamide adenine dinucleotide (NAD⁺) analogs were prepared starting from nicotinamide. The phosphodiester linkage was effectively assembled in 69–93% yields via condensation reaction between 2',3'-di-O-acetyl nicotinamide mononucleotide and alcohols in the presence of 2,4,6-triisopropylbenzenesulfonyl chloride. The analog b-nicotinamide ribose-5-(2-phenylethyl) phosphate showed beneficial effects on cell growth of model microorganisms.

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1. Introduction

Nicotinamide adenine dinucleotide (NAD⁺, Fig. 1) and its reduced form, NADH, serve as cofactors of many oxidoreductive en-zymes.^{[1](#page-5-0)} They are also functional molecules in many aspects of life. It has been established that $NAD⁺$ plays critical roles in calcium homeostasis, 2 2 cell proliferation, 3 aging, 4 4 apoptosis, 5 5 covalent protein modification, 6 6 gene expression, and regulation of numerous NAD⁺dependent non-oxidoreductive enzymes.[7](#page-5-0) Furthermore, recent studies showed that $NAD⁺$ could act as an immune modulator or

Figure 1. Structural function of $NAD⁺$ and its phosphodiester-type analogs.

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induce T cell death. 8 NAD⁺ uptake by mammalian cells has also been firmly recognized.^{[9](#page-5-0)} These findings have greatly enriched our understanding on the essential functions of pyridine nucleotides.^{[10](#page-5-0)} When NAD^+ functions as a coenzyme of dehydrogenase, the adenosine pocket and the NAR binding site are specially isolated.^{[11](#page-5-0)} This appeals to engineering of the adenosine binding pocket of dehydrogenase¹² to match elaborated NAD⁺ analogs in an approach as found in kinases research.^{[13](#page-5-0)} In such scenario, $NAD⁺$ analogs are of great potential to be developed for elucidation of complicated problems at the interface of chemistry and biology.

Although modifications on adenosine, ribose, PPi, and nicotinamide moiety of the NAD⁺ structure have been achieved, those with substantial changes of the NAD⁺ skeleton are limited.¹⁴ In particular, preparation of diversified phosphodiester-type $NAD⁺$ analogs (Fig. 1) remains challenging in terms of efficient coupling strategy and product purification. The simplest phosphodiester-type NAD^+ analog, ß-nicotinamide ribose-5'-methyl phosphate, was prepared using the N,N'-dicyclohexylcarbodiimide (DCC)/4-dimethylaminopyridine (DMAP) system.¹⁵ The reduced form of the analog acted as a biomimic cofactor binding to horse liver alcohol dehydrogenase and facilitated reduction of prochiral ketones in moderate vields.^{[16](#page-5-0)} Although the results were stimulating, no other phosphodiestertype NAD⁺ analogs have been documented so far. Because a methyl group could barely provoke tight interactions with protein, more complex structures are intriguing to mimic the adenosine moiety. Here, we designed novel phosphodiester-type NAD^+ analogs (Fig. 1), of which the adenine moiety, the ribose moiety, and the PPi linkage was replaced with other organic constituents, ether or alkane chains^{[17](#page-5-0)} and monophosphate functionality, respectively.

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We now report our efforts on preparation of these $NAD⁺$ analogs. We also present preliminary results of microbial cell growth and activity assay of dehydrogenases in the presence of a typical asymmetrical phosphodiester $NAD⁺$ analog.

2. Results and discussion

2.1. Synthetic plan

Construction of phosphatediesters including positive charged constituents remains challenge, although many strategies have been developed to assemble phosphodiester via either $P(III)^{18}$ $P(III)^{18}$ $P(III)^{18}$ or $P(V)^{19}$ $P(V)^{19}$ $P(V)^{19}$ reagents in nucleic acid chemistry.^{[20](#page-5-0)} $P(III)$ strategy required more steps such as oxidation, protection, and deprotection. This resulted in tedious purification work and breakdown of the anomeric C–N bond in NAD⁺ analog assembly. Thus, we focused on P(V) chemistry. Based on the retrosynthetic analysis, we planned our synthesis work by one of the two routes (Scheme 1). Intermolecular condensation reaction between nicotinamide mononucleotide (NMN, 4) and alcohols was designed as the key step in Route I. However, it was difficult to find an activation system to realize the coupling reaction, because NMN is liable to form an inner salt and has notorious solubility in most organic solvents. Moreover, breakdown of 4 via scission of the C–N bond was substantial in many experiments. According to Route II, we successfully prepared 6 in dichloromethane. However, attempts were in vain to assemble the C–N bond via glycosylation reaction with nicotinamide or its N-protected derivatives in the presence of Lewis acids. Therefore, these $NAD⁺$ analogs were prepared in a way similar to Route I with NMN derivatives (vide infra).

2.2. Synthesis

In the literatures, 4 has been prepared by selective hydrolysis 21 21 21 of NAD⁺ or chemical synthesis.²² To facilitate in-house preparation of various NAD⁺ analogs, we prepared 4 starting from commercially available nicotinamide according to a modified procedure and obtained a total isolated yield of 63%.[22c,23](#page-5-0) Briefly, nicotinamide was treated with 1,1,1,3,3,3-hexamethyldisilazane (HMDS) and TMSCl, then reacted with 1,2,3,5-tetra-O-acetylribofuranose (2) in the presence of trimethylsilyl trifluoromethanesulfonate (TMSOTf) to form 2',3',5'-tri-O-acetyl nicotinamide riboside via a glycosylation mechanism. The intermediate was deacetylated by methanolic ammonia to give nicotinamide ribose (NAR, 3), which was phosphorylated with OPCl₃ to afford 4. With NMN in hand, our attention turned to assemble the phosphodiester linkage en route to the $NAD⁺$ analogs.

The DCC/DMAP strategy was first applied to couple 4 with 2 phenylethanol (c), a conceptive adenosine analog, in various sol-vents.^{[15](#page-5-0)} It was found that long reaction time ($>$ 3 d) and large excess DCC were required to achieve good conversion, yet low yield and complicated byproducts were observed based on $31P$ NMR analysis. In the reaction, significant amount of 4 dimerized to form the symmetric pyrophosphate product. Reactions involved the hydroxyl groups at 2'- and 3'-positions of 4 also occurred. Consequently, it was difficult to recover β -nicotinamide ribose-5'-(2phenylethyl) phosphate 9c.

We then prepared 2',3'-di-O-acetyl nicotinamide mononucleotide (Ac₂NMN, 7) in around 62% overall yield started from nicotinamide (Scheme 2).^{[24](#page-5-0)} It was found that $AC₂NMN$ had reasonable solubility and high reactivity in solvents such as DMF,

Scheme 1. Retrosynthetic analysis for phosphodiester-type NAD⁺ analogs.

Scheme 2. Synthesis of phosphodiester-type NAD⁺ analogs. Reagents and conditions: (i) (a) TMSCl, HMDS, 120 °C, 8 h; (b) **2**, 1,2-dichloroethane, TMSOTf, 45 °C, 2 h; (c) NH₃/CH₃OH. -5 °C, 20-48 h; (ii) (a) POCl₃ (4 equiv), PO(OMe)₃, -5 to 0 °C, 12 h, 63% for four steps; (b) Ac₂O/pyridine (1:1), 0-5 °C, 24 h; (iii) (a) ROH (**a-n**), coupling reagent, DMF/pyridine (1:1) 25 °C, 6 h; (iv) 1 M NH₃/CH₃OH, -5 °C, 4 h, 96–98%. Data in the parentheses indicated the coupling yield.

DMSO, and pyridine, which were routinely used in coupling reactions. Moreover, blocking the free hydroxyl groups on the ribose ring also eliminated side reactions and improved the stability of the precursor.

The alcohols $(a-n)$ used in NAD⁺ analogs preparation were either commercially available (a–c) or obtained via Williamson ether synthesis $(d-n)$ using the corresponding phenol and the halide (Scheme 2).^{[25](#page-5-0)}

To attain these analogs, we evaluated commercially available condensation agents found in the preparations of DNA or RNA fragments (Fig. 2). Table 1 summarizes the results of the coupling reaction between Ac_2NMN (7) and 2-phenylethanol (c) with various condensation reagents using an anhydrous DMF/pyridine solvent system at room temperature. Conventional carbodiimides derivatives including DCC, N,N'-diisopropylcarbodiimide (DIC) and 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrochloride (EDCI) showed unsatisfactory results (Table 1, entries 1–3). Although about 100% conversion for 7 was achieved with these mediators, the product **8c** was less than 10% based on ${}^{31}P$ NMR analysis. Difficulties also occurred to recover 8c in the presence of excess byproducts and the coupling reagent derivatives. Acyl chlorides, commonly used for assembling of Hphosphonate linkage in nucleic acid chemistry, such as pivaloyl chloride (Piv-Cl) and 1-adamantanecarbonyl chloride (AC-Cl) were estimated.^{[20](#page-5-0)} Both Piv-Cl and AC-Cl facilitated the coupling reaction, yet the reaction was rather slow. It took over 10 d for Piv-Cl or AC-Cl to reach ³¹P NMR yields of 51% and 34%, respectively (Table 1, entries 4 and 5). Moreover, large excess 2-phenylethanol and the acyl chlorides were required. Chlorophosphates were also tested, including diphenyl phosphorochloridate (DPP-Cl) and bis(2-oxo-3-oxazolidinyl) phosphinic chloride (BOP-Cl), which are known to prompt asymmetrical pyrophosphate synthesis. Near quantitative conversion of 7 was achieved within 24 h in the presence of DPP-Cl or BOP-Cl (Table 1, entries 6 and 7).

Figure 2. Structures of condensation agents used in this work.

Table 1

^a Conditions: Ac₂NMN and 3 equiv of 2-phenylethanol were employed in the presence of a condensation agent in DMF/pyridine $(1:1)$ at 25 °C.

When $Ac₂NMN$ and 2-phenylethanol was treated with 2,4,6triisopropylbenzenesulfonyl chloride (TIPS-Cl), ^{31}P NMR analysis indicated that the singlet at 3.7 ppm of the starting material disappeared within $6 h$ and signals at 0.3 and -11.2 ppm, corresponding to the product and the pyrophosphate byproduct, respectively, evolved. A typical yield was about 93% (Table 1, entry 8) that was comparable to nucleotide or glycosyl phosphate diester synthesis using $P(III)$ agents.¹⁸ Seth et al. improved the yield of phosphodiester in nucleotide synthesis using 1H-tetrazole and TIPS-Cl. 26 However, dimerization of Ac₂NMN increased noticeably if 1H-tetrazole was applied.

Another agent with a similar structure to that of TIPS-Cl, 1-(2,4,6 triisopropylbenzenesulfonyl)-3-nitro-1H-1,2,4-triazole (TIPS-N) was estimated, however, the NMR yield was only 24% after 20 h (Table 1, entry 9).

As demonstrated herein, TIPS-Cl was the favored agent to prepare 8c. When it was applied for the coupling reaction with other alcohols, NAD⁺ analogs $8a-8n$ were obtained in isolated yields ranged from 69% to 93% [\(Scheme 2\)](#page-1-0). It was interesting to note that analogs 8j-8n, resulted from alcohols with an electron withdrawing or bulky group, were obtained in slightly lower yields. We reasoned that nucleophilic attack on the phosphorus atom of phosphate–sulfonate anhydride intermediate was less effective due to stronger repulsion forces between these alcohol substrates and the TIPS group.

2.3. Preliminary bioactivity assays

Preliminary tests were performed using the analog 9c for its effects on cell growth of model microorganisms. Compared to the control samples, Escherichia coli DH5a and Saccharomyces cerevisiae ATCC 26108 cells grew faster in the presence of 100 μ M 9c (Fig. 3). The maximal growth increments for E. coli and S. cerevisiae cells reached 16% and 52%, respectively. It is known that $NAD⁺$ or NAR can have beneficial effects on the culture of mammalian cells, because these compounds can change the metabolism of pyridine nucleotide cofactors.^{[9](#page-5-0)} The analog **9c** promoted microbial cell growth might also suggest a similar

Figure 3. Growth improvements of (A) E. coli DH5a and (B) S. cerevisiae ATCC 26108 in the presence of 100 $\upmu\mathrm{M}$ NAD⁺ analog **9c**. Cell growth was indicated using OD₆₀₀ data and normalized with the control cultures.

Enzymatic activity of alcohol dehydrogenase (ADH) from S. cerevisiae was also tested in the presence of analog 9c. Compared to the control sample, there was no discernible reaction rate change even the concentration of $9c$ was fivefold more than that of NAD⁺ (Fig. 4). Similar results were also observed in the case of lactate dehydrogenase (LDH) from Lactobacillus leichmanii (data not shown). These results indicated that the synthetic analogs were unlikely recognized by wild-type NAD⁺-dependent oxidoreductases. Extensive screening work is ongoing to find mutated enzymes from designed libraries.[12](#page-5-0)

Figure 4. Time course of ADH activity in the presence of 9c.

3. Conclusions

We have developed an effective and general strategy to prepare phosphodiester-type NAD^+ analogs starting from nicotinamide through coupling of Ac₂NMN with alcohols in the presence of TIPS-Cl. Preliminary study indicated that one representative analog 9c promoted microbial cell growth. We are now exploring the biological function with these compounds in a wide variety of chemical biology areas, and detailed results will be reported in due course.

4. Experimental

4.1. General

All reagents were analytical grade and obtained from commercial suppliers (ABCR, ACROS or Sigma). NMR spectra were measured with a Bruker DRX-400 spectrometer (400.3 MHz for ¹H, 100.6 MHz for ¹³C, 160.1 MHz for ³¹P, and 376.4 for ¹⁹F) at 298 K. HRMS was obtained on a Q-TOF-MS and operated with an electrospray source in positive ion mode. Optical density at 600 nm (OD_{600}) was recorded on JASCO V-530 UV-vis spectrophotometer. F_{254} thin-layer and silica gel (400 mesh) were purchased from Yantai Jiangyou Silica Co., Ltd., China. Octyl-functionalized silica gel was purchased from Sigma. Ion exchange resin (100–200 mesh) was purchased from the Chemical Plant of Nankai University, Tianjin, China. Bio-Gel P2 resin (45 µm) was obtained from Bio-Rad Laboratories, Inc. E. coli DH5a and S. cerevisiae ATCC 26108 were purchased from Beijing Ding Guo Biotech. Co., Ltd and Invitrogen Co., Ltd, respectively. ADH from S. cerevisiae (CAS No. 9031-72-5) and LDH from L. leichmanii (CAS No. 9028-36-8) were purchased from Sigma. All organic reactions were carried out under a nitrogen atmosphere.

4.2. Procedure for preparation of intermediate compounds

Adenosine analogs **d-n** in [Scheme 2](#page-1-0) were synthesized via Williamson ether synthesis using corresponding phenol and the halide according to known procedure (yield 80–92%)[.25](#page-5-0)

Ac₂NMN 7 was prepared starting from nicotinamide in five steps using a modified procedure of the literatures. $22c,23$

4.3. General procedure for preparation of NAD ⁺ analogs (9a–9n)

All NAD⁺ analogs were synthesized following the procedure described for $9c$. All purifications were performed at $25 °C$ except size exclusion chromatography $(4 °C)$.

4.3.1. β -Nicotinamide ribose-5-(2-phenylethyl) phosphate (**9c**). Ac₂NMN 7 (263 mg, 0.63 mmol) and 2-phenylethanol c (200 μ L, 1.67 mmol) were dissolved in pyridine/DMF (1:1, 25 mL). TIPS-Cl (604 mg, 2 mmol) was added and the suspension was stirred at room temperature for 6 h. After removing the solvents, the residue was taken by H₂O (15 mL), extracted with DCM (3×15 mL). The supernatant was purified by column chromatography on octylfunctionalized silica gel eluted with $CH₃OH/H₂O$ (30:1) to give the crude intermediate 8c (306 mg, 0.58 mmol). This was treated with 1 M NH₃/MeOH (0.2 mL) at -5 °C for 4 h. The solvent was evaporated and the residue was purified by column chromatography on anion resin (201 \times 2, HCO₂ form, H₂O). The corresponding fraction was pooled and lyophilized to give syrupy solid 9c (251 mg, 91%). 1 H NMR (400 MHz, D₂O) δ (ppm) 3.65 (m, 3H), 3.94 (m, 3H), 4.11 (t, J=2.2 Hz, 1H), 4.35 (s, 1H), 6.01 (d, J=5.4 Hz, 1H), 6.68 (m, 3H), 7.15 (m, 2H), 8.06 (t, J=7.6 Hz, 1H), 8.80 (d, J=8.0 Hz, 1H), 9.06 (d, J=6.0 Hz, 1H), 9.22 (s, 1H). ¹³C NMR (100 MHz, D₂O) δ (ppm) 62.3, 65.3, 66.7, 69.2, 73.4, 80.1, 89.6, 89.7, 102.2, 128.9, 130.8, 131.0, 131.6, 136.2, 141.2, 141.9, 144.7, 148.4, 167.0. ³¹P NMR (160 MHz, D₂O) δ (ppm) 0.2. HRMS (ESI) calcd for C₁₉H₂₃N₂O₈P [M+H]⁺ 439.1270; found: 439.1284.

4.3.2. β -Nicotinamide ribose-5-(1-butyl) phosphate (9a). Compound 9a (syrupy solid, 203 mg, 82%) was obtained from 7 (263 mg, 0.63 mmol) as described for the synthesis of $9c$, followed by size exclusion chromatography (polyacrylamide gel, $25 \text{ mM } NH_4HCO_3$). ¹H NMR (400 MHz, D₂O) δ (ppm) 0.73 (t, J=7.2 Hz, 2H), 1.19 (m, 2H), 1.44 (m, 2H), 3.73 (q, 2H), 4.01 (d, J=5.3 Hz, 1H), 4.18 (d, J=10.3 Hz, 1H), 4.3 (s, 1H), 4.40 (t, J=5.1 Hz, 1H), 4.49 (s, 1H), 6.08 (d, J=4.3 Hz, 1H), 8.15 (t, J=6.9 Hz, 1H), 8.86 (d, J=7.9 Hz, 1H), 9.14 (d, J=6.0 Hz, 1H), 9.34 (s, 1H). ¹³C NMR (100 MHz, D₂O) δ (ppm) 15.4, 20.8, 34.4, 66.8, 68.7, 73.5, 80.2, 89.7, 102.3, 130.9, 136.4, 142.2, 144.9, 148.4, 168.2. 31P NMR (160 MHz, D_2O) δ (ppm) 0.5. HRMS (ESI) calcd for $C_{15}H_{23}N_2O_8P$ $[M+H]$ ⁺ 391.1270; found: 391.1277.

4.3.3. β -Nicotinamide ribose-5-[2-(1-adamantane)ethyl] phosphate (9b). Compound 9b (syrupy solid, 271 mg, 87%) was obtained from 7 (263 mg, 0.63 mmol) as described for the synthesis of 9c, followed by size exclusion chromatography (polyacrylamide gel, 25 mM NH₄HCO₃). ¹H NMR (400 MHz, D₂O) δ (ppm) 1.20 (t $J=7.4$ Hz, 2H), 1.28 (d, $J=2.0$ Hz, 6H), 1.39–1.51 (m, 6H), 1.69 (s, 3H), 3.74 (q, 2H), 4.01 (ddd, J=1.9, 4.9, 12.0 Hz, 1H), 4.17 (ddd, J=2.2, 4.0, 12.1 Hz, 1H), 4.28 (s, 1H), 4.43 (s, 1H), 4.48 (s, 1H), 6.12 (d, J=5.0 Hz, 1H), 8.21 (t, J=7.2 Hz, 1H), 8.92 (d, J=7.9 Hz, 1H), 9.19 (d, J=5.8 Hz, 1H), 9.37 (s, 1H). ¹³C NMR (100 MHz, D₂O) δ (ppm) 30.8, 33.7, 39.1, 44.5, 46.5, 65.0, 66.8, 73.3, 80.2, 89.7, 102.5, 131.1, 136.5, 142.3, 145.2, 148.7, 167.8. ³¹P NMR (160 MHz, D₂O) δ (ppm) 0.6. HRMS (ESI) calcd for C₂₃H₃₃N₂O₈P [M+H]⁺ 497.2053; found: 497.2061.

4.3.4. b-Nicotinamide ribose-5-[5-(m-tolyloxy)pentyl] phosphate (**9d**). Compound **9d** (syrupy solid, 266 mg, 83%) was obtained from **7** (263 mg, 0.63 mmol) as described for the synthesis of $9c$. ¹H NMR $(400$ MHz, D₂O) δ (ppm) 1.34 (m, 2H), 1.56 (m, 4H), 2.10 (s, 3H), 3.77 $(m, 4H)$, 4.00 (ddd, J=2.5, 4.8, 10.4 Hz, 1H), 4.14 (ddd, J=2.0, 4.0, 12.0 Hz, 1H), 4.27 (m, 1H), 4.32 (t, J=4.8 Hz, 1H), 4.44 (s, 1H), 5.99 (d, $J=5.1$ Hz, 1H), 6.53 (t, J=7.2 Hz, 2H), 6.65 (d, J=7.5 Hz, 1H), 7.01 (t, $J=7.8$ Hz, 1H), 8.07 (t, $J=6.4$ Hz, 1H), 8.73 (d, $J=8.1$ Hz, 1H), 9.07 (d, $J=6.2$ Hz, 1H), 9.20 (s, 1H). ¹³C NMR (100 MHz, D₂O) δ (ppm) 22.9, 24.3, 30.4, 31.9, 66.8, 68.7, 70.5, 73.1, 80.1, 89.6, 102.3, 114.0, 117.7, 124.5, 130.9, 132.1, 136.2, 142.2, 142.7, 144.7, 148.3, 160.5, 167.7. 31P NMR (160 MHz, D_2O) δ (ppm) 0.6. HRMS (ESI) calcd for $C_{23}H_{31}N_2O_9P$ [M+H]⁺ 511.1845; found: 511.1839.

4.3.5. β -Nicotinamide ribose-5-{2-[2-(m-tolyloxy)ethoxy]ethyl} phosphate (9e). Compound 9e (syrupy solid, 267 mg, 83%) was obtained from **7** (263 mg, 0.63 mmol) as described for the synthesis of $9c$. $^1\mathrm{H}$ NMR (400 MHz, D_2O) δ (ppm) 2.04 (s, 3H), 3.59 (t, J=4.2 Hz, 2H), 3.67 $(t, J=7.2$ Hz, 2H), 3.88 (m, 4H), 3.97 (ddd, J=1.9, 4.8, 12.1 Hz, 1H), 4.15 $(\text{ddd}, \text{J} = 1.8, 3.4, 12.1 \text{ Hz}, 1\text{H}), 4.21 \text{ (m, 1H)}, 4.25 \text{ (t, J} = 5.1 \text{ Hz}, 1\text{H}), 4.34$ $(s, 1H)$, 5.86 (d, J=5.2 Hz, 1H), 6.43 (m, 2H), 6.57 (d, J=7.5 Hz, 1H), 6.93 (t, J=8.3 Hz, 1H), 7.99 (t, J=7.5 Hz, 1H), 8.63 (d, J=8.1 Hz, 1H), 9.00 (d, J=6.2 Hz, 1H), 9.08 (s, 1H), ¹³C NMR (100 MHz, D₂O) δ (ppm) 66.8, 67.6, 69.4, 71.6, 72.7, 73.0, 80.1, 89.5, 102.2, 117.5, 124.6, 129.2, 130.9, 132.0, 136.0, 141.9, 142.6, 144.6, 148.1, 160.1, 162.8, 167.6. 31P NMR (160 MHz, D_2O) δ (ppm) 0.3. HRMS (ESI) calcd for $C_{22}H_{29}N_2O_{10}P$ [M+H]⁺ 513.1638; found: 513.1632.

4.3.6. β -Nicotinamide ribose-5-{2-[2-(2-phenoxy)ethoxy]ethyl} phosphate (9f). Compound 9f (syrupy solid, 242 mg, 86%) was obtained from **7** (263 mg, 0.63 mmol) as described for the synthesis of $9c$. 1 H NMR (400 MHz, D_2O) δ (ppm) 3.64 (t, J=4.3 Hz, 2H), 3.74 (t, J=4.2 Hz, 2H), 3.91 (m, 2H), 3.99 (m, 3H), 4.15 (dd, J=2.5, 5.5 Hz, 2H), 4.25 (d, $J=5.3$ Hz, 1H), 4.29 (d, J=2.6 Hz, 1H), 4.38 (s, 1H), 5.93 (d, J=5.4 Hz, 1H), 6.73 (d, J=8.2 Hz, 2H), 6.82 (t, J=7.4 Hz, 1H), 7.13 (t, J=7.7 Hz, 1H), 8.05 $(t, J=6.7$ Hz, 1H), 8.69 (d, J=8.1 Hz, 1H), 9.06 (d, J=6.2 Hz, 1H), 9.15 (s, 1H). ¹³C NMR (100 MHz, D₂O) δ (ppm) 66.9, 67.6, 69.6, 71.7, 72.7, 73.2, 80.2, 89.6, 102.3, 117.0, 124.0, 129.4, 131.0, 132.3, 136.2, 142.1, 144.8, 148.3, 160.2, 167.9. ³¹P NMR (160 MHz, D₂O) δ (ppm) 0.3. HRMS (ESI) calcd for $C_{21}H_{27}N_2O_{10}P$ [M+H]⁺ 449.1482; found: 449.1493.

4.3.7. β -Nicotinamide ribose-5-(5-phenoxypentyl) phosphate (9g). Compound 9g (syrupy solid, 265 mg, 85%) was obtained from 7 (263 mg, 0.63 mmol) as described for the synthesis of **9c.** ¹H NMR (400 MHz, D₂O) δ (ppm) 1.30 (m, 2H), 1.48 (m, 2H), 1.54 (m, 2H), 3.71 (q, 2H), 3.75 (t, J=6.4 Hz, 2H), 3.95 (ddd, J=2.2, 4.9, 12.0 Hz, 1H), 4.10 (ddd, J=2.4, 4.2, 12.0 Hz, 1H), 4.22 (m, 1H), 4.28 (t, J=5.0 Hz, 1H), 4.39 (s, 1H), 5.96 (d, J=5.2 Hz, 1H), 6.68 (d, J=8.1 Hz, 2H), 6.77 (t, J=7.4 Hz, 1H), 7.09 (t, J=8.4 Hz, 2H), 8.02 (t, J=7.8 Hz, 1H), 8.69 (d, J=8.1 Hz, 1H), 9.02 (d, J=6.3 Hz, 1H), 9.16 (s, 1H). ¹³C NMR (100 MHz, D₂O) δ (ppm) 24.2, 30.4, 31.8, 31.9, 66.8, 68.7, 70.5, 73.0, 80.1, 89.4, 102.3, 117.0, 123.7, 130.9, 132.2, 136.2, 142.1, 144.7, 148.3, 160.4, 162.7, 167.8. ³¹P NMR (160 MHz, D₂O) δ (ppm) 0.5. HRMS (ESI) calcd for $C_{22}H_{29}N_2O_9P$ [M+H]⁺ 497.1689; found: 497.1680.

4.3.8. β -Nicotinamide ribose-5-{2-[2-(2-methoxyphenoxy)ethoxy]ethyl} phosphate (9h). Compound 9h (syrupy solid, 297 mg, 86%) was obtained from 7 (263 mg, 0.63 mmol) as similar procedure described for the synthesis of $9c$ (anion resin, aqueous 5% HCO₂NH₄), followed by column chromatography on octyl-functionalized silica gel with CH₃OH/H₂O (30:1) as eluent. ¹H NMR (400 MHz, D₂O) δ (ppm) 3.61 (s, 3H), 3.64 (t, J=2.7 Hz, 2H), 3.75 (t, J=4.2 Hz, 2H), 3.87–4.08 (m, 5H), 4.20 (ddd, J=2.1, 4.5, 12.2 Hz, 1H), 4.24 (dd, J=2.3, 4.7 Hz, 1H), 4.26 (t, $J=5.1$ Hz, 1H), 4.38 (s, 1H), 5.88 (d, $J=5.3$ Hz, 1H), 6.77–6.91 (m, 4H), 8.03 (t, J=7.7 Hz, 1H), 8.68 (d, J=8.0 Hz, 1H), 9.02 (d, J=6.2 Hz, 1H), 9.14 $(s, 1H)$. ¹³C NMR (100 MHz, D₂O) δ (ppm) 57.8, 66.9, 67.7, 70.2, 71.6, 72.7, 73.2, 80.1, 89.5, 102.2, 114.4, 115.5, 123.8, 124.2, 130.9, 136.1, 141.8, 144.7, 148.1, 149.3, 150.4, 162.8. ³¹P NMR (160 MHz, D₂O) δ (ppm) 0.3. HRMS (ESI) calcd for $C_{22}H_{29}N_2O_{11}P[M+H]^+$ 551.1407; found: 551.1402.

4.3.9. β -Nicotinamide ribose-5-[5-(2-methoxyphenoxy)pentyl] phosphate (9i). Compound 9i (syrupy solid, 277 mg, 82%) was obtained

from 7 (263 mg, 0.63 mmol) as similar procedure described for the synthesis of **9c** (anion resin, aqueous 5% HCO₂NH₄), followed by column chromatography on octyl-functionalized silica gel with CH₃OH/H₂O (30:1) as eluent. ¹H NMR (400 MHz, D₂O) δ (ppm) 1.37 (q, 2H), 1.54 (m, 2H), 1.63 (m, 2H), 3.66 (s, 3H), 3.76 (q, 2H), 3.84 (t, J=6.2 Hz, 2H), 4.01 (dd, J=4.5, 12.2 Hz, 1H), 4.16 (dd, J=4.1, 12.0 Hz, 1H), 4.27 (m, 1H), 4.34 (t, J=5.0 Hz, 1H), 4.45 (s, 1H), 5.99 (d, J=5.2 Hz, 1H), 6.81 (m, 4H), 8.08 (t, J=7.3 Hz, 1H), 8.72 (d, J=8.0 Hz, 1H), 9.06 $(d, J=6.2 \text{ Hz}, 1\text{ H}), 9.22 \text{ (s, 1H)}.$ ¹³C NMR (100 MHz, D₂O) δ (ppm) 24.2, 30.3, 31.9, 57.9, 66.8, 68.7, 71.2, 73.1, 80.1, 89.6, 114.5, 115.7, 123.9, 124.0, 130.6, 130.9, 136.2, 142.2, 148.3, 149.7, 150.7, 162.7, 167.8. ³¹P NMR (160 MHz, D_2O) δ (ppm) 0.6. HRMS (ESI) calcd for $C_{23}H_{31}N_2O_{10}P$ [M+H]⁺ 527.1795; found: 527.1802.

4.3.10. β -Nicotinamide ribose-5-{2-[2-(3-fluorophenoxy)ethoxy]ethyl} phosphate $(9i)$. Compound $9i$ (syrupy solid, 234 mg, 72%) was obtained from 7 (263 mg, 0.63 mmol) as similar procedure described for the synthesis of $9c$ (anion resin, aqueous 5% HCO₂H). ¹H NMR $(400$ MHz, D₂O) δ (ppm) 3.65 (s, 2H), 3.75 (t, J=3.8 Hz, 2H), 3.91 (m, 2H), 3.99 (m, 3H), 4.17 (dd, J=3.9, 11.9 Hz, 1H), 4.26 (m, 1H), 4.33 (m, 1H), 4.40 (m, 1H), 5.97 (d, J=5.2 Hz, 1H), 6.56 (m, 3H), 7.12 (m, 1H), 8.10 (t, J=8.1 Hz, 1H), 8.73 (d, J=8.7 Hz, 1H), 9.08 (d, J=6.2 Hz, 1H), 9.21 (s, 1H). ¹³C NMR (100 MHz, D₂O) δ (ppm) 66.9, 67.6, 69.9, 71.5, 72.7, 72.8, 73.2, 80.2, 89.5, 89.6, 102.3, 104.4, 104.7, 110.3, 110.6, 112.9, 131.0, 133.1, 133.2, 136.2, 142.1, 144.8, 148.3, 161.6, 167.8. 31P NMR (160 MHz, D₂O) δ (ppm) 0.3. ¹⁹F NMR (376 MHz, D₂O) δ (ppm) -111.5 . HRMS (ESI) calcd for C₂₁H₂₆FN₂O₁₀P [M+H]⁺ 517.1406; found: 517.1397.

4.3.11. β-Nicotinamide ribose-5-[5-(3-fluorophenoxy)pentyl] phosphate ($9k$). Compound $9k$ (syrupy solid, 217 mg, 67%) was obtained from 7 (263 mg, 0.63 mmol) as similar procedure described for the synthesis of **9c** (anion resin, aqueous 5% HCO₂H). ¹H NMR $(400$ MHz, $D_2O)$ δ (ppm) 1.37 (q, 2H), 1.57 (m, 2H), 1.64 (m, 2H), 3.78 $(q, 2H)$, 3.87 (t, J=6.2 Hz, 2H), 4.05 (dd, J=4.5, 11.6 Hz, 1H), 4.20 (dd, $J=4.3$, 12.0 Hz, 1H), 4.31 (m, 1H), 4.38 (t, $J=5.0$ Hz, 1H), 4.48 (s, 1H), 6.05 (d, $I=5.3$ Hz, 1H), 6.60 (m, 3H), 7.15 (m, 1H), 8.14 (t, $I=7.2$ Hz, 1H), 8.78 (d, J=8.0 Hz, 1H), 9.12 (d, J=6.1 Hz, 1H), 9.28 (s, 1H). ¹³C NMR (100 MHz, D₂O) δ (ppm) 66.8, 68.8, 71.0, 73.2, 80.2, 89.6, 89.7, 102.4, 104.5, 104.7, 110.1, 110.3, 113.1, 131.0, 133.1, 133.2, 136.3, 142.2, 144.8, 148.4, 162.4, 167.5. ³¹P NMR (160 MHz, D₂O) δ (ppm) 0.6. ¹⁹F NMR (376 MHz, D_2O) δ (ppm) -111.7. HRMS (ESI) calcd for $C_{22}H_{28}FN_{2}O_{9}P$ [M+H]⁺ 515.1595; found: 515.1592.

4.3.12. β -Nicotinamide ribose-5-{2-[2-(2-chlorophenoxy)ethoxy]ethyl} phosphate (91). Compound 91 (syrupy solid, 231 mg, 69%) was obtained from 7 (263 mg, 0.63 mmol) as similar procedure described for the synthesis of $9c$ (anion resin, aqueous 5% HCO₂H). ¹H NMR $(400$ MHz, D₂O) δ (ppm) 3.61 (t, J=4.1 Hz, 2H), 3.70 (t, J=4.2 Hz, 2H), 3.86 (m, 2H), 3.99 (m, 3H), 4.13 (ddd, $J=2.4$, 3.9 , 11.9 Hz, 2H), 4.18 (m, 1H), 4.23 (t, $I=5.1$ Hz, 1H), 4.32 (t, $I=2.4$ Hz, 1H), 5.87 (d, $I=5.2$ Hz, 1H), 6.70 (t, J=7.6 Hz, 2H), 6.79 (d, J=8.3 Hz, 1H), 7.03 (m, 2H), 8.00 (t, J=6.4 Hz, 1H), 8.63 (d, J=8.1 Hz, 1H), 8.99 (d, J=6.3 Hz, 1H), 9.10 (s, 1H). ¹³C NMR (100 MHz, D₂O) δ (ppm) 66.8, 67.7, 67.8, 71.0, 71.5, 72.8, 73.0, 80.1, 89.3, 89.4, 102.2, 116.6, 124.0, 124.6, 130.8, 130.9, 132.6, 136.0, 142.0, 144.7, 148.2, 155.7, 162.7, 167.6. ³¹P NMR (160 MHz, D₂O) δ (ppm) 0.2. HRMS (ESI) calcd for C₂₁H₂₆ClN₂O₁₀P [M+H]⁺ 533.1092; found: 533.1092.

4.3.13. β-Nicotinamide ribose-5-[5-(2-chlorophenoxy) pentyl] phosphate ($9m$). Compound $9m$ (syrupy solid, 224 mg, 68%) was obtained from 7 (263 mg, 0.63 mmol) as similar procedure described for the synthesis of $9c$ (anion resin, aqueous 5% HCO₂H). ¹H NMR (400 MHz, D_2O) δ (ppm) 1.42 (q, 2H), 1.56 (m, 2H), 1.66 (m, 2H), 3.80 (q, 2H), 3.93 (t, J=5.6 Hz, 2H), 4.00 (ddd, J=2.3, 4.6, 12.1 Hz, 1H), 4.16 (ddd, J=2.0, 4.1, 12.0 Hz 1H), 4.29 (m, 1H), 4.36 (m, 1H), 4.46 (m,

1H), 6.03 (d, J=5.1 Hz, 1H), 6.81 (t, J=7.7 Hz, 1H), 6.92 (d, J=8.3 Hz, 1H), 7.13 (t, J=7.7 Hz, 1H), 7.2 (d, J=9.9 Hz, 1H), 8.11 (t, J=7.7 Hz, 1H), 8.73 (d, J=8.1 Hz, 1H), 9.11 (d, J=6.2 Hz, 1H), 9.27 (s, 1H). ¹³C NMR $(100$ MHz, D₂O) δ (ppm) 24.3, 30.3, 31.9, 66.8, 68.8, 71.8, 73.1, 80.2, 89.5, 102.4, 116.9, 124.5, 130.8, 131.0, 132.6, 136.2, 142.3, 144.8, 148.3, 156.0, 167.7. ³¹P NMR (160 MHz, D₂O) δ (ppm) 0.6. HRMS (ESI) calcd For $C_{22}H_{28}C/N_2O_9P$ [M+H]⁺ 533.1119; found: 533.1111.

4.3.14. b-Nicotinamide ribose-5-{2-[2-(N-benzoyl-3-aminophenoxy) ethoxy]ethyl} phosphate ($9n$). Compound $9n$ (syrupy solid, 287 mg, 74%) was obtained from 7 (263 mg, 0.63 mmol) as similar procedure described for the synthesis of 9c (octyl-functionalized silica gel, $CH₃OH/H₂O$, 30:1), followed by size exclusion chromatography (polyacrylamide gel, 25 mM NH₄HCO₃). ¹H NMR (400 MHz, D₂O) δ (ppm) 3.69 (m, 2H), 3.79 (m, 2H), 4.02 (m, 4H), 4.08 (m, 1H), 4.17 $(d, J=11.9 \text{ Hz } 1\text{H})$, 4.30 (m, 2H), 4.39 (s, 1H), 5.89 (d, J = 5.2 Hz, 1H), 6.64 (d, J=8.3 Hz, 1H), 6.94 (d, J=8.0 Hz, 2H), 6.99 (s, 1H), 7.18 (t, J=8.1 Hz, 2H), 7.43 (t, J=7.5 Hz, 2H), 7.53 (d, J=7.4 Hz, 1H), 7.72 $(d, J=8.0$ Hz, 2H), 8.06 $(t, J=7.2$ Hz, 1H), 8.70 $(d, J=8.0$ Hz, 1H), 9.06 $(d, J=6.1 \text{ Hz}, 1\text{ H})$, 9.13 (s, 1H). ¹³C NMR (100 MHz, D₂O) δ (ppm) 66.9, 67.7, 69.7, 71.6, 72.8, 73.2, 80.1, 89.6, 102.3, 110.4, 113.7, 117.0, 129.8, 130.9, 131.3, 132.6, 134.9, 136.0, 136.2, 140.9, 142.1, 144.6, 148.2, 160.7, 171.5. 31 P NMR (160 MHz, D₂O) δ (ppm) 0.2. HRMS (ESI) calcd for $C_{28}H_{32}N_3O_{11}P$ [M+H]⁺ 618.1853; found: 618.1857.

4.4. Preliminary tests of the effects of 9c on microbial cell growth

The detailed procedure for cell culture in the presence of $NAD⁺$ analog 9c was as follows: 55 mL sterilized Luria–Bertani medium (peptone 10 g/L, yeast extract 5 g/L, NaCl 10 g/L, pH 7.2) supplemented with 100 μ M analog **9c** was inoculated with 8-h-old preculture of E. coli DH5 α cells (500 µL), and cultured at 37 °C, 200 rpm. The control sample was run in the absence of analog 9c. Aliquots were withdrawn every 2 h and $OD₆₀₀$ was recorded on a UV–vis spectrophotometer.

For yeast S. cerevisiae ATCC 26108, 50 mL sterilized YEPD medium (glucose 20 g/L, yeast extract 10 g/L, peptone 20 g/L, pH 6.2) supplemented with 100 μ M analog 9c was inoculated with 18-hold preculture (250 μ L), and cultured at 30 °C, 200 rpm. The control sample was run in the absence of analog 9c. Aliquots were withdrawn every 5 h and OD_{600} was recorded on a UV-vis spectrophotometer.

4.5. Activity assay of ADH in the presence of analog 9c

To a solution of ethanol (10 mM), NAD⁺ (5 mM), and analog **9c** (25 mM) in Tris–HCl buffer (50 mM, pH 8.8) was added ADH (0.002 U). The mixture was vortexed and transferred into a colorimetric cuvette. Time course measurement was recorded for 90 s at 340 nm on a UV-vis spectrophotometer at 25° C. Activity of LDH was similarly tested.

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Supplementary data

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References and notes

- 1. Stryer, L. Biochemistry; W.H. Freeman: New York, NY, 1995.
- 2. Guse, A. H.; Gu, X. F.; Zhang, L.-R.; Weber, K.; Krämer, E.; Yang, Z.-J.; Jin, H.-W.; Li, Q.; Carrier, L.; Zhang, L.-H. J. Biol. Chem. 2005, 280, 15952.
- 3. Bruzzone, S.; Flora, A. D.; Usai, C.; Graeff, R.; Lee, H. C. Biochem. J. 2003, 375, 395.
- 4. Blasco, M. A. Nat. Rev. Genet. 2005, 6, 611.
- 5. Gendron, M. C.; Schrantz, N.; Metivier, D.; Kroemer, G.; Maciorowska, Z.; Sureau, F.; Koester, S.; Petit, P. X. Biochem. J. 2001, 353, 357.
- 6. Mandir, A. S.; Simbulan-Rosenthal, C. M.; Poitras, M. F.; Lumpkin, J. R.; Dawson, V. L.; Smulson, M. E.; Dawson, T. M. J. Neurochem. 2002, 83, 186.
- 7. (a) Girolamo, M. D.; Dani, N.; Stilla, A.; Corda, D. FEBS J. 2005, 272, 4565; (b) Haince, J.-F.; Kozlov, S.; Dawson, V. L.; Dawson, T. M.; Hendzel, M. J.; Lavin, M. F.; Poirier, G. G. J. Biol. Chem. 2007, 282, 16441; (c) Sauve, A. A.; Schramm, V. L. Biochemistry 2003, 42, 9249; (d) Michan, S.; Sinclair, D. Biochem. J. 2007, 404, 1; (e) Nakano, T.; Matsushima-Hibiya, Y.; Yamamoto, M.; Enomoto, S.; Matsumoto, Y.; Totsuka, Y.; Watanabe, M.; Sugimura, T.; Wakabayashi, K. Proc. Natl. Acad. Sci. U.S.A. 2006, 103, 13652; (f) Culver, G. M.; McCraith, S. M.; Consaul, S. A.; Stanford, D. R.; Phizicky, E. M. J. Biol. Chem. 1997, 272, 13203; (g) Berger, F.; Lau, C.; Ziegler, M. Proc. Natl. Acad. Sci. U.S.A. 2007, 104, 3765.
- 8. (a) Song, E.-K.; Lee, Y.-R.; Yu, H.-N.; Kim, U.-H.; Rah, S.-Y.; Park, K.-H.; Shim, I.-K.; Lee, S.-J.; Park, Y.-M.; Chung, W.-G.; Kim, J.-S.; Han, M.-K. Biochem. Biophys. Res. Commun. 2008, 367, 156; (b) Seman, M.; Adriouch, S.; Scheuplein, F.; Krebs, C.; Freese, D.; Glowacki, G.; Deterre, P.; Haag, F.; Koch-Nolte, F. Immunity 2003, 19, 571.
- 9. (a) Billington, R. A.; Travelli, C.; Ercolano, E.; Galli, U.; Roman, C. B.; Grolla, A. A.; Canonico, P. L.; Condorelli, F.; Genazzani, A. A. J. Biol. Chem. 2008, 283, 6367; (b) Yang, T.; Chan, N. Y.-K.; Sauve, A. A. J. Med. Chem. 2007, 50, 6458; (c) Verderio, C.; Bruzzone, S.; Zocchi, E.; Fedele, E.; Schenk, U.; De Flora, A.; Matteoli, M. J. Neurochem. 2001, 78, 646.
- 10. (a) Koch-Nolte, F.; Haag, F.; Guse, A. H.; Lund, F.; Ziegler, M. Science 2009, 2, 1; (b) Lin, H. Org. Biomol. Chem. 2007, 5, 2541.
- 11. Tracewell, C. A.; Arnold, F. H. Curr. Opin. Chem. Biol. 2009, 13, 3.
- 12. Wang, J.-X.; Zhang, S.-F.; Tan, H.-D.; Zhao, Z.-B. J. Microbiol. Methods 2007, 71, 225.
- 13. (a) Habelhah, H.; Shah, K.; Huang, L.; Burlingame, A. L.; Shokat, K. M.; Ronai, Z. J. Biol. Chem. 2001, 276, 18090; (b) Elphick, L. M.; Lee, S. E.; Child, E. S.; Prasad, A.; Pignocchi, C.; Thibaudeau, S.; Anderson, A. A.; Bonnac, L.; Gouverneur, V.; Mann, D. J. ChemBioChem 2009, 10, 1519.
- 14. Woenckhaus, C.; Jeck, R. In Coenzymes and Cofactors, Pyridine Nucleotide Coenzyme; Dolphin, D., Avramovic, O., Poulson, R., Eds.; John Wiley & Sons: New York, NY, 1987; Vol. 2A, pp 450–568.
- 15. Lo, H. C.; Leiva, C.; Buriez, O.; Kerr, J. B.; Olmstead, M. M.; Fish, R. H. Inorg. Chem. 2001, 40, 6705.
- 16. Lo, H. C.; Fish, R. H. Angew. Chem., Int. Ed. 2002, 41, 478.
- 17. Xu, J.-F.; Yang, Z.-J.; Dammermann, W.; Zhang, L.-R.; Guse, A. H.; Zhang, L.-H. Med. Chem. 2006, 49, 5501.
- 18. Nikolaev, A. V.; Botvinko, I. V.; Ross, A. J. Carbohydr. Res. 2007, 342, 297.
- 19. Stawinski, J.; Kraszewski, A. Acc. Chem. Res. 2002, 35, 952.
- 20. (a) Winqvist, A.; Strömberg, R. Eur. J. Org. Chem. 2008, 1705; (b) Zain, R.; Stawiński, J. J. Org. Chem. 1996, 61, 6617.
- 21. (a) Liu, R. H.; Visscher, J. Nucleosides Nucleotides Nucleic Acids 1994, 13, 1215; (b) Sankyo Co. Ltd., J.P. Patent 70,000,948, 1970.
- 22. (a) Walt, D. R.; Findeis, M. A.; Rios-Mercadillo, V. M.; Auge, J.; Whitesides, G. M. J. Am. Chem. Soc. 1984, 106, 234; (b) Walt, D. R.; Rios-Mercadillo, V. M.; Auge, J.; Whitesides, G. M. J. Am. Chem. Soc. 1980, 102, 7805; (c) Lee, J.; Churchil, H.; Choi, W.-B.; Lynch, J. E.; Roberts, F. E.; Volante, R. P.; Reider, P. J. Chem. Commun. 1999, 729.
- 23. Franchetti, P.; Pasqualini, M.; Petrelli, R.; Ricciutelli, M.; Vita, P.; Cappellacci, L. Bioorg. Med. Chem. Lett. 2004, 14, 4655.
- 24. Graham, S. M.; Macaya, D. J.; Sengupta, R. N.; Turner, K. B. Org. Lett. 2004, 6, 233.
- 25. (a) Lazny, R.; Nodzewska, A.; Klosowski, P. Tetrahedron 2004, 60, 121; (b) Ashton, P. R.; Ballardini, R.; Balzani, V.; Constable, E. C.; Credi, A.; Kocian, O.; Langford, S. J.; Preece, J. A.; Prodi, L.; Schofield, E. R.; Spencer, N.; Stoddart, J. F.; Wenger, S. Chem.-Eur. J. 1998, 4, 2413; (c) Ahmed, S. A.; Tanaka, M. J. Org. Chem. 2006, 71, 9884.
- 26. Seth, A. K.; Jay, E. Nucleic Acids Res. 1980, 8, 5445.